

Declassified in Part - Sanitized Copy Approved for Release 2012/04/10: CIA-RDP78-03424A000500040008-6 Eng-7160-103 Director of Logistics Director of Communications 00 9084 Request for Contract Amendment, RD-107, Task Order 11; 1 2 FEB 1950 25X1 1. Under the terms of ND-107, Task Order 11, is developing a water activated 25X1 battery for this Office. Due to difficulties in obtaining materials for this development early in the program, this task has extended beyond the original schedule and will now be completed approximately 5 months after acceptance of this amendment. 2. A request for additional funds to cover this extension of time has been received from the contractor in a letter dated 19 January 1960, a copy of which is attached. In this letter two preposals for completing this task are given. This office has carefully examined these proposals and requests that proposal "A" be accepted less payment of the \$660.00 fee. The reason for this increase in funds and time extension stems from the fact that the contractor was unable to procure silver clad 25X1 magnesium. It was necessary, therefore, for the to provide a means of producing a suitable material. It is 25X1 our opinion that this does not constitute a change in scope and the fee of \$660.00 is not justified. 3. Therefore, it is requested that Task Order 11 under RB-107 be amended to authorise these additional costs requested by the contractor, and extend the termination date 5 months from acceptance of this amendment. Attached is Requisition No. NEB 60-375 indicating that the allotment to be charged is 9/7900-50-095. Funds in the amount of \$8,251.00 have been encumbered for this purpose. This tack order is Agency sterile, but the equipment resulting therefrom is UNCLARSIFIED. 25X1 The project engineer for this program is 25X1 Extension REV DATE \_\_ ORIG COMP 033 CM 56 TYPE 02 25X1 ORIG CLASS -5 PACES 9 REV CLASS 22 NEXT REV 20/0 AUTH: HR 78-2 Acting ACCECHENIALY (1) Contractor's letter dated 19 January 1960 (2) Requisition No. MEB 60-375 Coordination: Distribution: Original and 1 - Addressee w/attachments R+D Subject File w/o attachments OC-A w/o attachments w/o attachments OC-T w/o attachments MSB w/o attachments OC-E Chrono w/o attachments OC-E/R+D/EP:CWS:maf (8 Feb. 60) Declassified in Part - Sanitized Copy Approved for Release 2012/04/10 : CIA-RDP78-03424A000500040008-6

bee:		
	January 19, 1960	
Subject:	Program Extension	
Reference:	Water-Activated Battery	
Dear Sir:		
of this of of further proposals	rming our telephone conversation between your  fice on January 6, 1960,  in the interesting the effort and results thus far obtained, is hereby submitting to for an extension of the present requisition. It is to be noted in the state of the present of a special time with this effort here.	wo he
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estimated on a CPFF basis	of \$6,151. Propos	al "B" is also
described in Exhibit "C" as Proposal "A" with the excep	nd is essentially	identical to r of end items
and laboratory effort.	beton of one numbe	1 Of Cha forms
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In order to properly plan our these proposals to sixty days. A	manpower commitme written extension:	may be affected thereafter
if agreeable to both parties.		
If there are any questions or	if further inform	etion is needed, please do no
hesitate to contact the undersigned		more to meeting, product as in
	Ve	ery truly yours,

CJP/sak Enclosures -

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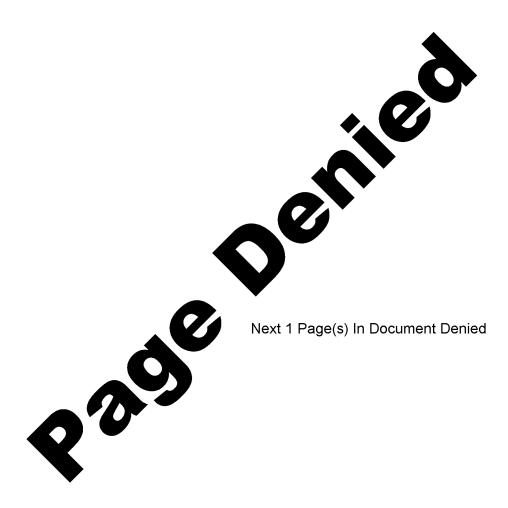


EXHIBIT "C"

Proposed Work

for

The Extension of

#### Water Activated Battery Development

25X1

# I. Objectives of Present Contract

#### Phase 1:

To demonstrate the feasibility of a large model, chemically rechargeable, magnesium-silver chloride battery, activated by a 3 per cent salt solution at room temperature and delivering an average current of 3 A at 12 V for at least 60 minutes. The voltage regulation at a maximum load current of 5.3 A shall equal or surpass the stability obtained previously.

#### Phase 2:

To demonstrate reasonably satisfactory performance of this battery at environmental temperatures of -40°C and 40°C, using a variety of electrolytes such as 3 per cent salt solution, tap water and others. It is recognized, however, that such performance cannot be expected to match that at room temperature.

#### Phase 3:

To fabricate and deliver six battery cases and thirty complete sets of chemical recharges for field testing purposes, including instructions on the proper handling, activation, deactivation, disposal and temperature control.

#### II. Achievements During 1959

Our achievements to date can be summarized as follows:

- 1. A study was made of the basic reactions and discharge mechanisms of this battery with special emphasis on the processes contributing to the internal generation of heat.
- 2. A new technique for producing a suitable bond between silver and magnesium foils was successfully developed.
- 3. This technique was extended to include other battery elements, leading to a substantial reduction in the number of battery components and a corresponding simplification in operating procedure.
- 4. It was found that other components (spacers) could be eliminated with only a small sacrifice in capacity, leading to an additional simplification in operating procedure.
- 5. The battery was designed, constructed and successfully tested at environmental temperatures of +25°C.
- 6. Exploratory tests indicate, that activation and operation of the battery at -40°C, with only a small sacrifice in capacity, is possible if suitable liquid salt solutions are added to the cold battery case. The minimum permissible temperature of the liquid electrolyte varies between -40°C and +20°C depending on the salt solution used.
- 7. A new type of battery cooling, based on the evaporation of a small amount of water applied externally to specially designed battery panels, has been developed, which promises the successful operation of the battery at +40°C without the need for additional accessories and/or increases in size and weight.

Achievements 1-5 are fully described in our bimonthly reports Nos. 1 and 2 while achievements 6 and 7 are the subject of report No. 3 which is in preparation. In summary it can be said that the entire phase 1 and a substantial part of phase 2 have been completed.

### III. Reasons for Underestimating the Scope of this Program

The most important modification of this program became apparent at its very beginning. At that time it was found, contrary to expectation, that the procurement of silver clad magnesium, although available in small sample form, was not possible in moderately large quantities. Consequently, as described in report No. 1, a new task for the development of a new technique for producing a suitable bond between silver and magnesium foils had to be added to the program. The performance of this task, achievement No. 2, required two months of intensive work.

The desirability of another modification became apparent only after it was found that the assembly and disassembly of a chemical charge required considerable manual dexterity and the use of some simple tools. This further development lead to a drastic reduction in battery compounds, described in report No. 2 (achievements 3 and 4).

The time required to carry out these tasks (achievements 2, 3 and 4) made completion of phase 3 impossible on the present contract.

## IV. Effort Required to Complete the Modified Program

Since most technical problems (Phase 1) of this program were solved during 1959, there is no doubt that the successful completion of the program can be virtually assured. For completion of phase 2 the

exploratory testing of a few additional, potentially interesting electrolytes both at low and high temperatures will be required. This should be supplemented by some representative full scale battery tests at both temperature extremes.

The commercial non-availability of silver clad magnesium not only increased the magnitudes of the development effort of Phase 1, but it also greatly increased the effort required for the fabrication program of Phase 3, which is to be initiated after the completion of Phase 2. Consequently, it is estimated that the completion of the program will require an intensive effort on fabrication and assembly during a four to six month period following the resumption of work. During this period about three months of technician's time, one month of a man in the shop and two weeks of supervision will be required.

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